(J. Chem. Soc., Perkin Trans. 1, 1987, 187)

10-Thia-anthracenes. Part 3. A Re-examination of the Reaction of 9-Phenylthioxanthylium Salt and Phenyl-lithium.

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Reaction of 9-phenylthioxanthylium salt with phenyl-lithium afforded eight compounds, 9-phenyl-, 9,9-diphenyl-, 3,9-diphenyl-, 3,9,9-triphenyl-, and 4,9,9-triphenylthioxanthene, 9,9'-diphenyl-9,9'-bithioxanthenyl, thioxanthone, and 9,9'-diphenyldithioxanth-9-yl peroxide. Their structures were determined by comparison with the authentic samples. A radical mechanism contributed to the reaction of 9-phenylthioxanthylium salt with phenyl-lithium.

(J. Chem. Soc., Perkin Trans. 1, 1987, 1211)

Reactions of Xanthinium N(7)-Ylides with Olefinic Dipolarophiles.

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The stereo- and regio-chemical aspects of the reactions of xanthinium N(7)-methylides with olefinic dipolarophiles were elucidated. The reactions of ylides with N-phenylmaleimide afforded stereoselective endo adducts, and with acrylates and acrylonitrile afforded stereo- and regioselective 6-endo adducts. On the other hand, the reactions of ylides with trans-olefins afforded mixtures of two stereoisomers, the 8-endo-7-exo and 8-exo-7-endo pyrrolo(2,1-f) purine derivatives. Stereochemistry of the adducts was elucidated by 1H n.m.r. and X-ray analysis. The stereoselectivity of the adducts was governed by the balance of steric and electronic effects. The ylides reacted in Z-form in all reactions investigated.

(J. Chem. Soc., Perkin Trans. 1, 1987, 2531)

An Unexpected Double Cycloaddition of (1,2,4)Triazolo(1,5-a) pyrimidine N-Ylide with Activated Acetylenes and Alkenes.

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The reaction of 5,7-dimethyl(1,2,4)triazolo(1,5-a)pyrimidinio-3-phenacylide with activated acety-lenes gave 1:2 adducts of ylide-R-C=C-R. The structures of the products were determined as 3,3a-dihydropyrazolo(1,5-c)pyrimidine derivatives by hydrolysis, 1H and ^{13}C n. m. r., and X-ray crystal-lography. Molecular orbital calculations (ab initio) of the model compounds were performed in order to elucidate the mechanism for the formation of the 1:2 adducts. The results of the calculations suggested that an intermediate, 1:1 adduct would be less reactive than the starting ylide. Easy formation of the 1:2 adducts can be explained by an equilibrium between the 1:1 adducts and the starting materials.